

Davangere



University

Shivagangothri, Davangere- 577007

PROGRAM /COURSE STRUCTURE AND SYLLABUS
of
PHYSICS

as per the Choice Based Credit System (CBCS) designed in accordance with


Learning Outcomes-Based Curriculum Framework (LOCF) of


National Education Policy (NEP) 2020

Bachelor of Science (B.Sc. Physics)

w.e.f.

Academic Year 2021-22 and onward


DEAN
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Chairman
Board of Studies
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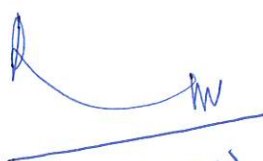
B. Sc. V and VI Semester

Program Outcomes	
1	Disciplinary knowledge
2	Communication Skills
3	Critical thinking, Reflective thinking, Analytical reasoning, Scientific reasoning
4	Problem-solving
5	Research-related skills
6	Cooperation/ Teamwork/ Leadership readiness/Qualities
7	Information / Digital literacy / Modern Tool Usage
8	Environment and Sustainability
9	Multicultural competence
10	Multi-Disciplinary
11	Moral and ethical awareness / Reasoning
12	Lifelong learning / Self Directed Learning

SEMESTER - V										
Category	Course code	Title of the Paper	Marks			Teaching hours/week			Credit	Duration of exams (Hours)
			IA	SEE	Total	L	T	P		
DSC	21BSC5C5PHY5L	Classical Mechanics and Quantum Mechanics- I	40	60	100	4	-	-	4	2
	21BSC5C5PHY5P	Classical Mechanics and Quantum Mechanics- I (Practical)	25	25	50	-	-	4	2	3
	21BSC5C5PHY6L	Elements of Atomic, Molecular and Laser Physics	40	60	100	4	-	-	4	2
	21BSC5C5PHY6P	Elements of Atomic, Molecular & Laser Physics (Practical)	25	25	50	-	-	4	2	3
Total Marks					300	Semester Credits			12	



Registrar
Davangere University
Shivagangothri, Davangere



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Program Name	B. Sc. Physics	Semester	V
Course Title	Classical Mechanics and Quantum Mechanics- I		
Course Code:	21BSC5C5PHY5L	No. of Credits	04
Contact hours	60 Hours	Duration of SEA/Exam	2 Hours
Formative Assessment Marks	40	Summative Assessment Marks	60

Course Pre-requisite(s):

Course Outcomes (COs): After the successful completion of the course, the student will be able to

- Identify the failure of classical physics at the microscopic level.
- Find the relationship between the normalization of a wave function and the ability to correctly calculate expectation values or probability densities.
- Explain the minimum uncertainty of measuring both observables on any quantum state.
- Describe the time-dependent and time-independent Schrödinger equation for simple potentials like for instance one-dimensional potential well and Harmonic oscillator.
- Apply Hermitian operators, their eigenvalues and eigenvectors to find various commutation and uncertainty relations.

Contents	60 Hours
<p align="center">Unit-I</p> <p>Introduction to Newtonian Mechanics: Frames of references, inertial and non-inertial frames. Newton's laws of motion, Mechanics of a particle, Conservation of linear momentum, Angular momentum and torque, conservation of angular momentum, work done by a force, conservative force and conservative energy.</p> <p>Lagrangian formulation: Constraints, Holonomic constraints, non-holonomic constraints, Scleronomic and Rheonomic constraints. Generalized coordinates, degrees of freedom, Principle of virtual work, D'Alembert's principle, Lagrange equations. Newton's equation of motion from Lagrange equations, Simple pendulum, Atwood's machine and linear harmonic oscillator.</p>	15 Hours
<p align="center">Unit-II</p> <p>Variational principle: Hamilton's principle, Deduction of Hamilton's principle, Lagrange's equation of motion from Hamilton's principle, Hamilton's principle for non-holonomic systems.</p> <p>Hamiltonian Mechanics: The Hamiltonian of a system, Hamilton's equations of motion, Hamilton's equations from variational principle, Integrals of Hamilton's equations, energy integrals, Canonical Transformations, Poisson Brackets, fundamental properties and equations of motion in Poisson Brackets.</p>	15 Hours
<p align="center">Unit-III</p> <p>Introduction to Quantum Mechanics: Brief discussion on failure of classical physics to explain black body radiation, Photoelectric effect, Compton effect, stability of atoms and spectra of atoms. Compton scattering: Expression for Compton shift (With derivation).</p> <p>Matter Waves: de Broglie hypothesis of matter waves, Electron microscope, Wave description of particles by wave packets, Group and Phase velocities and relation between them, Experimental evidence for matter waves: Davisson- Germer experiment, G.P Thomson's experiment and its significance.</p> <p>Heisenberg uncertainty principle: Elementary proof of Heisenberg's relation between momentum and position, energy and time, angular momentum and angular position, illustration of uncertainty principle by Gamma ray microscope thought experiment. Consequences of the uncertainty relations: Diffraction of electrons at a single slit, Non-existence of electron in the nucleus.</p>	15 Hours

<p style="text-align: center;">Unit-IV</p> <p>Foundation of Quantum Mechanics: Probabilistic interpretation of the wave function - normalization and orthogonality of wave functions, Admissibility conditions on a wave function, Schrödinger equation: equation of motion of matter waves - Schrodinger wave equation for a free particle in one and three-dimension, time-dependent and time-independent wave equations, Probability current density, equation of continuity and its physical significance, Postulates of quantum mechanics: States as normalized wave functions. Dynamical variables as linear Hermitian operators (position, momentum, angular momentum, and energy as examples). Expectation values of operators and their time evolution. Ehrenfest theorem (no derivation), Commutator brackets- Simultaneous Eigen functions, Commutator bracket using position, momentum and angular momentum operators. Particle in a one-dimensional infinite potential well (derivation), particle in a finite potential well (qualitative).</p>	15 Hours
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Pedagogy: Lecture/ PPT/ Videos/ Animations/ Role Plays/ Think-Pair-Share/ Predict-Observe-Explain/ Demonstration/ Concept mapping/ Case Studies examples/ Tutorial/ Activity/ Flipped Classroom/ Jigsaw/ Field based Learning/ Project Based Learning/ Mini Projects/ Hobby Projects/ Forum Theatre/ Dance/ Problem Based Learning/ Game Based Learning/ Group Discussion/ Collaborative Learning/ Experiential Learning / Self Directed Learning etc.

Formative Assessment for Theory	
Assessment Occasion/ type	Marks
Internal Assessment	10
Project/Assignment	10
Activities	10
Presentation (Oral/Poster)	10
Total	40 Marks
<i>Formative Assessment as per UNIVERSITY guidelines are compulsory.</i>	

References	
1	H. Goldstein, C.P. Poole, J.L. Safko, Classical Mechanics, 3rd Edn. 2002, Pearson Education.
2	Dieter Strauch, Classical Mechanics: An introduction, 2009, Springer.
3	G. Aruldas, Classical Mechanics, 2008, Prentice-Hall of India Private limited, New Delhi.
4	R. G. Takwale and T. S. Puranik, Classical Mechanics, 1989, Tata McGraw Hill, New Delhi.
5	Arthur Beiser, Concepts of Modern Physics, 2009, McGraw-Hill,
6	Serway and Jewett, Physics for Scientists and Engineers with Modern Physics, 9 th edition, 2014, Cengage Learning.
7	E. H. Wichman, Quantum Physics, Berkeley Physics Course Vol. 4. 2008, Tata McGraw-Hill Co.
8	P M Mathews and K Venkatesan, A Textbook of Quantum Mechanics, 2017, Tata McGraw Hill Publication.
9	Ajoy Ghatak, S. Lokanathan, Quantum Mechanics: Theory and Applications, 2004, Springer Publication.
10	C. L. Arora and P.S. Hemne, Physics for Degree Students B.Sc., Third Year, 1st edition, 2014, S. Chand & Company Pvt. Ltd.

Course Title	Classical Mechanics and Quantum Mechanics-I (Practical)	Practical Credits	02
Course Code	21BSC5C5PHY5P	Contact Hours	4 Hours
Formative Assessment	25 Marks	Summative Assessment	25 Marks

Practical Content

Lab experiments: (Minimum 8 experiments from the following).

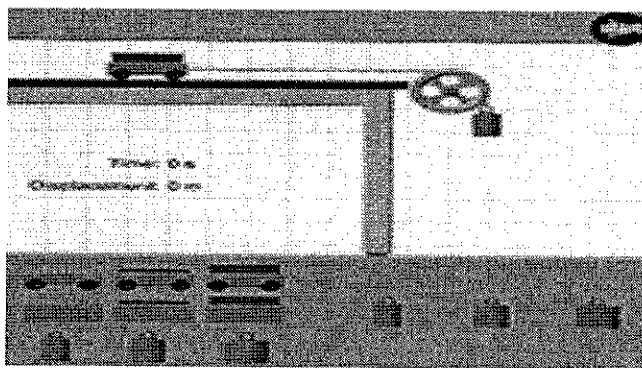
- 1) To determine 'g', the acceleration due to gravity, at a given place, from the $L - T^2$ graph, for a simple pendulum.
- 2) Studying the effect of mass of the bob on the time period of the simple pendulum.
[Hint: With the same experimental set-up, take a few bobs of different materials (different masses) but of same size. Keep the length of the pendulum same for each case. Starting from a small angular displacement of about 10° find out, in each case, the time period of the pendulum, using bobs of different masses. Does the time period depend on the mass of the pendulum bob? If yes, then see the order in which the change occurs. If not, then do you see an additional reason to use the pendulum as a time measuring device.]
- 3) Studying the effect of amplitude of oscillation on the time period of the simple pendulum.
[Hint: With the same experimental set-up, keep the mass of the bob and length of the pendulum fixed. For measuring the angular amplitude, make a large protractor on the cardboard and have a scale marked on an arc from 0° to 90° in units of 5° . Fix it on the edge of a table by two drawing pins such that its 0° - line coincides with the suspension thread of the pendulum at rest. Start the pendulum oscillating with a very large angular amplitude (say 70°) and find the time period T of the pendulum. Change the amplitude of oscillation of the bob in small steps of 5° or 10° and determine the time period in each case till the amplitude becomes small (say 5°). Draw a graph between angular amplitude and T . How does the time period of the pendulum change with the amplitude of oscillation? How much does the value of T for $A = 10^\circ$ differ from that for $A = 50^\circ$ from the graph you have drawn? Find at what amplitude of oscillation, the time period begins to vary? Determine the limit for the pendulum when it ceases to be a simple pendulum.]
- 4) Determine the acceleration of gravity is to use an Atwood's machine.
- 5) Study the conservation of energy and momentum using projectile motion.
- 6) Verification of the Principle of Conservation of Linear Momentum
- 7) Determination of Planck constant and work function of the material of the cathode using Photo-electric cell.
- 8) To study the spectral characteristics of a photo-voltaic cell (Solar cell).
- 9) Determination of electron charge 'e' by Millikan's Oil drop experiment.
- 10) To study the characteristics of solar cell.
- 11) To find the value of e/m for an electron by Thomson's method using bar magnets.
- 12) To determine the value of e/m for an electron by magnetron method.
- 13) To study the tunnelling in Tunnel Diode using I-V characteristics.
- 14) Determination of quantum efficiency of Photodiode.
- 15) A code in C/C++/Scilab to find the first seven eigen states and eigen functions of Linear Harmonic Oscillator by solving the Schrödinger equation.
- 16) A code in C/C++/Scilab to plot and analyse the wave functions for particle in an infinite potential well.

Pedagogy: Demonstration/Experiential Learning / Self Directed Learning etc.

Formative Assessment for Practical	
Assessment Occasion/ type	Marks
Practical Internal Assessment	10
Record	10
Attendance	05
Total	25 Marks
<i>Formative Assessment as per UNIVERSITY guidelines are compulsory.</i>	

References	
1	C. L. Arora, B. Sc. Practical Physics. 2001, S. Chand Publishing.
2	Singh, Harnam, and P. S. Hemne. B. Sc. Practical Physics, 2000, S. Chand Publishing.
3	Squires, Gordon Leslie. Practical Physics. 2001, Cambridge university press.
4	Dr Neetu Agrawal Physics,, Scilab Manual for CC-XI: Quantum Mechanics & Applications (32221501) Daulat Ram College, August 2023, University of Delhi,.
5	M. C. Jain, Companion for Quantum Mechanics, 2019, Scilab Textbook,.
6	D. Ganguly, R. R, A. Mitra, S. Barua, Computational Quantum Mechanics using Scilab, 2021, BIT Mesra.
7	Worsnop B L and Flint H T, 2017, Advanced Practical Physics for Students.

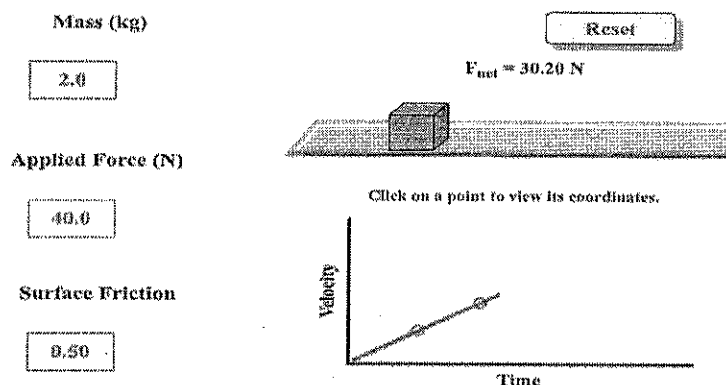
Activities



Atwood's Machine

Everyone is fascinated by pulleys. In this Interactive, learners will attach two objects together by a string and stretch the string over a pulley. Both an Atwood's machine and a modified Atwood's machine can be created and studied. Change the amount of mass on either object, introduce friction forces, and measure distance and time in order to calculate the acceleration.

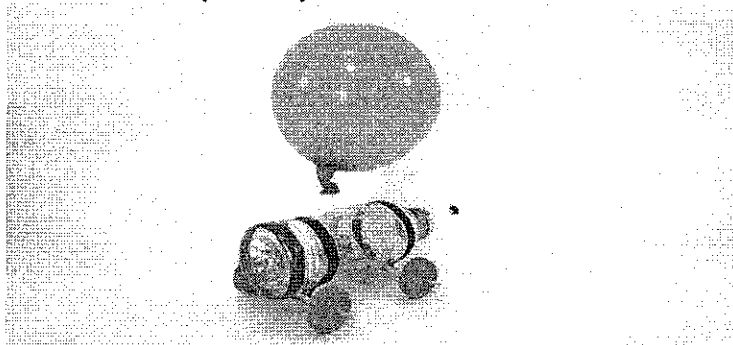
Newton's Laws of Motion



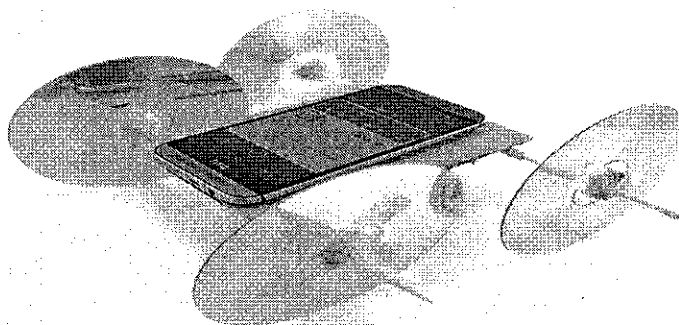
Force

When forces are unbalanced, objects accelerate. But what factors affect the amount of acceleration? This Interactive allows learners to investigate a variety of factors that affect the acceleration of a box pushed across a surface. The amount of applied force, the mass, and the friction can be altered. A plot of velocity as a function of time can be used to determine the acceleration.

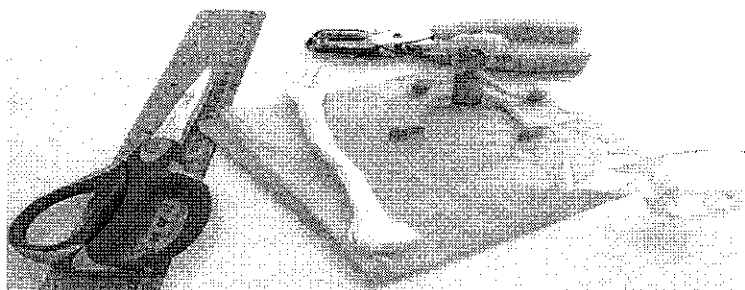
In the [Balloon Car Lesson Plan](#), students build and explore balloon-powered cars. This lesson focuses mostly on energy, but it also demonstrates Newton's laws of motion. Guidance is provided for talking specifically about the third law of motion. *Question:* how does the air escaping the balloon relate to Newton's third law of motion? Does the car continue to coast after the balloon is deflated? Why or why not?



Most of the activities and lessons below *focus* on one or two of the laws of motion. The Build a Balloon Car activity specifically **talks about all three of Newton's laws of motion** students can observe when building and experimenting with a simple balloon-powered car. This is an accessible hands-on activity that uses recycled materials and balloons for a fun combined engineering design project and physics experiment. The activity can be used with a wide range of grade levels to introduce and demonstrate the laws of motion. See the "Digging Deeper" section for a straightforward discussion of how each law of motion can be identified in the balloon car activity. (For a related lesson plan, see Balloon Car Lesson Plan, which is NGSS-aligned for middle school and focuses on the third law of motion.). In the Push Harder — Newton's Second Law, students build their own cars using craft materials and get hands-on exploring Newton's second law of motion and the equation "force equals mass times acceleration" ($F=ma$). Options for gathering real-time data include using a mobile phone and a sensor app or using a meter stick and a stopwatch. *Questions:* What is the relationship between force, mass, and acceleration? As force increases, what happens to acceleration?



In the Skydive Into Forces, students make parachutes and then investigate how they work to slow down a falling object. As students investigate the forces that are involved, educators can introduce Newton's second law of motion and how different forces change the resulting speed of a falling object. *Questions:* What forces help slow down the speed of a falling object? How does a parachute help slow the fall?



2

Both standard cameras (DSLRs, phone cameras) and our scientific cameras work on the principle of photoelectric effect to produce an image from light, involving the use of **photodetectors** and **sensor pixels**. **Prepare a report on the working of digital camera.**

3	<p>Demonstration of Heisenberg uncertainty principle in the context of diffraction at a single slit: The uncertainty in the momentum Δp_x correspond to the angular spread of principal maxima θ. Then, $\Delta p_x = \sin \theta \cdot p$ where p is the momentum of the incident photon.</p> <p>Conduct the diffraction at a slit experiment virtually using the following link https://www.walter-fendt.de/html5/phen/singleslit_en.htm</p> <ol style="list-style-type: none"> 1. Measure the angular spread (θ) for different slit widths (Δx) for given wavelength of the incident photon. 2. Determine the momentum of the incident photon using $p = \frac{h}{\lambda}$ 3. Create a line of best fit through the points in the plot $\frac{1}{\Delta p_x}$ against Δx and find its slope. <p>How this exercise is related to Heisenberg Uncertainty principle, Make a report of the observations.</p>
4	<p>Virtual lab to demonstrate Photoelectric effect using <i>Value@Amritha</i>: Conduct the virtual experiment using the following link https://vlab.amrita.edu/?sub=1&brch=195&sim=840&cnt=1</p> <ol style="list-style-type: none"> 1. Determine the minimum frequency required to have Photoelectric effect for an EM radiation, when incident on a zinc metal surface. 2. Determine the target material if the threshold frequency of EM radiation is 5.5×10^{15} Hz in a particular photoelectric experimental set up. 3. Determine the maximum kinetic energy of photo-electrons emitted from a Zinc metal surface, if the incident frequency is 3×10^{15} Hz. 4. What should be the stopping potential for photoelectrons if the target Material used is Platinum and incident frequency is 2×10^{15} Hz? Make a report of the calculations.
5	<p>Visualization of wave packets using Physlet@Quantum Physics: The concept of group velocity and phase velocity of a wave packet can be studied using this link https://www.compadre.org/PQP/quantum-need/section5_9.cfm Students can take up the exercises using the link which is as follows https://www.compadre.org/PQP/quantum-need/prob5_11.cfm Six different classical wave packets are shown in the animations. Which of the wave packets have a phase velocity that is: greater than / less than / equal to the group velocity? Make a report of the observations.</p>
6	<p>Superposition of eigen states in an infinite one - dimensional potential well using QuVis (Quantum Mechanics Visualization Project): Construct different possible states by considering the first three eigen states and study the variation of probability density with position. Take the challenges after understanding the simulation and submit the report. The link is as follows https://www.standrews.ac.uk/physics/quvis/simulations_html5/sims/SuperpositionStates/SuperpositionStates.html</p>
7	<p>Determination of expectation values of position, momentum for a particle in a an infinite one - dimensional potential well using Physlet@Quantum Physics: The link to the visualization tool for the calculation is as follows https://www.compadre.org/PQP/quantum-theory/prob10_3.cfm A particle is in a one-dimensional box of length $L = 1$. The states shown are normalized. The results of the integrals that give $\langle x \rangle$ and $\langle x^2 \rangle$ and $\langle p \rangle$ and $\langle p^2 \rangle$. You may vary n from 1 to 10.</p> <ol style="list-style-type: none"> a) What do you notice about the values of $\langle x \rangle$ and $\langle x^2 \rangle$ as you vary n? b) What do you think $\langle x^2 \rangle$ should become in the limit of $n \rightarrow \infty$? Why? c) What do you notice about the values of $\langle p \rangle$ and $\langle p^2 \rangle$ as you vary n? <p>Make a report of the calculations.</p>

8	<p>Determination of expectation values for a particle in a one-dimensional harmonic oscillator using Physlet@Quantum Physics:</p> <p>The link to the visualization tool for the calculation is as follows https://www.compadre.org/PQP/quantum-theory/prob12_2.cfm</p> <p>A particle is in a one-dimensional harmonic oscillator potential ($\hbar = 2m = 1$; $\omega = k = 2$). The states shown are normalized. Shown are ψ and the results of the integrals that give $\langle x \rangle$ and $\langle x^2 \rangle$ and $\langle p \rangle$ and $\langle p^2 \rangle$. Vary n from 1 to 10.</p> <ol style="list-style-type: none"> What do you notice about how $\langle x \rangle$ and $\langle x^2 \rangle$ and $\langle p \rangle$ and $\langle p^2 \rangle$ change? Calculate $\Delta x \cdot \Delta p$ for $n = 0$. What do you notice considering $\hbar = 1$? What is E_n? How does this agree with or disagree with the standard case for the harmonic oscillator? How much average kinetic and potential energies are in an arbitrary energy state? <p>Make a report of the calculations.</p>
9	<p>Calculate uncertainties of position and momentum for a particle in a box using Physlet@Quantum Physics:</p> <p>The link to the visualization tool for the calculation is as follows https://www.compadre.org/PQP/quantum-theory/prob6_3.cfm</p> <p>A particle is in a one-dimensional box of length $L = 1$. The states shown are normalized. The results of the integrals that give $\langle x \rangle$ and $\langle x^2 \rangle$, and $\langle p \rangle$ and $\langle p^2 \rangle$. You may vary n from 1 to 10.</p> <ol style="list-style-type: none"> For $n = 1$, what are Δx and Δp? For $n = 10$, what are Δx and Δp?

Program Name	B. Sc. Physics	Semester	V
Course Title	Elements of Atomic, Molecular and Laser Physics		
Course Code:	21BSC5C5PHY6L	No. of Credits	04
Contact hours	60 Hours	Duration of SEA/Exam	2 Hours
Formative Assessment Marks	40	Summative Assessment Marks	60

Course Pre-requisite (s): PUC Science Knowledge

Course Outcomes (COs): After the completion of the course, the student will be able to

- Describe atomic properties using basic atomic models.
- Interpret atomic spectra of elements using vector atom model.
- Interpret molecular spectra of compounds using basics of molecular physics.
- Explain laser systems and their applications in various fields.

Contents	60 Hours
<p style="text-align: center;">Unit I</p> <p>Basic Atomic models: Thomson's atomic model; Rutherford atomic model – Model, Theory of alpha particle scattering, Rutherford scattering formula; Bohr atomic model – postulates, Derivation of expression for radius, total energy of electron; Origin of the spectral lines; Spectral series of hydrogen atom; Effect of nuclear motion on atomic spectra - derivation; Ritz combination principle; Correspondence principle; Critical potentials – critical potential, excitation potential and ionisation potential, Atomic excitation and its types, Franck-Hertz experiment; Sommerfeld's atomic model – model, Derivation of condition for allowed elliptical orbits.</p> <p style="text-align: right;">12 Hours</p> <hr/> <p>Activities:</p> <p style="text-align: right;">03 Hours</p> <ol style="list-style-type: none"> 1. Students to estimate radii of orbits and energies of electron in case of hydrogen atom in different orbits and plot the graph of radii / energy versus principal quantum number 'n'. Analyse the nature of the graph and draw the inferences. 2. Students to search critical, excitation and ionisation potentials of different elements and plot the graph of critical / excitation / ionisation potentials versus atomic number/mass number/neutron number of element. Analyse the nature of the graph and draw the inferences. 	15 Hours

<p style="text-align: center;">Unit II</p> <p>Vector atomic model and optical spectra: Vector atom model – model fundamentals, spatial quantisation, spinning electron; Quantum numbers associated with vector atomic model; Coupling schemes – L-S and j-j schemes; Pauli's exclusion principle; Magnetic dipole moment due to orbital motion of electron – derivation; Magnetic dipole moment due to spin motion of electron; Lande g-factor and its calculation for different states; Stern-Gerlach experiment – Experimental arrangement and Principle; Fine structure of spectral lines with examples; Spin-orbit coupling/Spin-Orbit Interaction – qualitative; Optical spectra – spectral terms, spectral notations, selection rules, intensity rules; Fine structure of the sodium D-line; Zeeman effect: Types, Experimental study and classical theory of normal Zeeman effect, Zeeman shift expression (no derivation), examples; Stark effect: Experimental study, Types and examples.</p> <p style="text-align: right;">12 Hours</p> <hr/> <p>Activities: 03 Hours</p> <ol style="list-style-type: none"> Students to couple a p-state and s-state electron via L-S and j-j coupling schemes for a system with two electrons and construct vector diagrams for each resultant. Analyse the coupling results and draw the inferences. Students to estimate magnetic dipole moment due to orbital motion of electron for different states $^2P_{1/2}$, $^2P_{3/2}$, $^2P_{5/2}$, $^2P_{7/2}$, $^2P_{9/2}$ and $^2P_{11/2}$ and plot the graph of dipole moment versus total orbital angular momentum 'J'. Analyse the nature of the graph and draw the inferences. 	15 Hours
<p style="text-align: center;">Unit III</p> <p>Molecular Physics: Types of molecules based on their moment of inertia; Types of molecular motions and energies; Born-Oppenheimer approximation; Origin of molecular spectra; Nature of molecular spectra; Theory of rigid rotator – energy levels and spectrum, Qualitative discussion on Non-rigid rotator and centrifugal distortion; Theory of vibrating molecule as a simple harmonic oscillator – energy levels and spectrum; Electronic spectra of molecules – fluorescence and phosphorescence; Raman effect – Stoke's and anti-Stoke's lines, characteristics of Raman spectra, classical and quantum approaches, Experimental study of Raman effect; Applications of Raman effect.</p> <p style="text-align: right;">12 Hours</p> <hr/> <p>Activities: 03 Hours</p> <ol style="list-style-type: none"> Students to estimate energy of rigid diatomic molecules CO, HCl and plot the graph of rotational energy versus rotational quantum number 'J'. Analyse the nature of the graph and draw the inferences. Also students study the effect of isotopes on rotational energies. 	15 Hours

2. Students to estimate energy of harmonic vibrating molecules CO, HCl and plot the graph of vibrational energy versus vibrational quantum number 'v'. Analyse the nature of the graph and draw the inferences.	
<p style="text-align: center;">Unit IV</p> <p>Laser Physics : Ordinary light versus laser light; Characteristics of laser light; Interaction of radiation with matter - Induced absorption, spontaneous emission and stimulated emission with mention of rate equations; Einstein's A and B coefficients – Derivation of relation between Einstein's coefficients and radiation energy density; Possibility of amplification of light; Population inversion; Methods of pumping; Metastable states; Requisites of laser – energy source, active medium and laser cavity; Difference between Three level and four level lasers with examples; Types of lasers with examples; Construction and Working principle of Ruby Laser and He-Ne Laser; Application of lasers (qualitative) in science & research, isotope separation, communication, fusion, medicine, industry, war and space.</p> <p style="text-align: right;">12 Hours</p> <hr/> <p>Activities:</p> <p style="text-align: right;">03 Hours</p> <ol style="list-style-type: none"> Students to search different lasers used in medical field (ex: eye surgery, endoscopy, dentistry etc.), list their parameters and analyse the need of these parameters for specific application, and draw the inferences. Students also make the presentation of the study. Students to search different lasers used in defence field (ex: range finding, laser weapon, etc.), list their parameters and analyse the need of these parameters for specific application, and draw the inferences. Students also make the presentation of the study. 	15 Hours

Pedagogy: Lecture/ PPT/ Videos/ Animations/ Role Plays/ Think-Pair-Share/ Predict-Observe-Explain/ Demonstration/ Concept mapping/ Case Studies examples/ Tutorial/ Activity/ Flipped Classroom/ Jigsaw/ Field based Learning/ Project Based Learning/ Mini Projects/ Hobby Projects/ Forum Theatre/ Dance/ Problem Based Learning/ Game Based Learning/ Group Discussion/ Collaborative Learning/ Experiential Learning / Self Directed Learning etc.

Formative Assessment for Theory	
Assessment Occasion/ type	Marks
Internal Assessment	10
Project/Assignment	10
Activities	10
Presentation (Oral/Poster)	10
Total	40 Marks
Formative Assessment as per UNIVERSITY guidelines are compulsory.	

References:

1	R. Murugesan, K. Sivaprakash, Modern Physics, Revised Edition, 2009, S. Chand & Company Ltd.
2	Raj Kumar, Atomic & Molecular spectra: Laser, , Revised Edition, 2008, Kedar Nath Ram Nath Publishers, Meerut.
3	S. N. Ghoshal, Atomic Physics, Revised Edition, 2013, S. Chand & Company Ltd.
4	S.P. Kuila, Concepts of Atomic Physics, First Edition, 2018, New Central Book Agency (P) Ltd.
5	Arthur Beiser Concepts of Modern Physics, Seventh Edition, 2015, Shobhit Mahajan, S. Rai Choudhury, 2002, McGraw-Hill.
6	C. N. Banwell and E.M. McCash, Fundamentals of Molecular Spectroscopy, Fourth Edition, 2008, Tata McGraw-Hill Publishers.
7	Gupta, Kumar and Sharma, Elements of Spectroscopy – Atomic, Molecular and Laser Physics, 2016, Pragati Publications.

Course Title	Elements of Atomic, Molecular and Laser Physics (Practical)	Practical Credits	02
Course Code	21BSC5C5PHY6P	Contact Hours	04 Hours
Formative Assessment	25 Marks	Summative Assessment	25 Marks
Practical Content			
Lab experiments: (Minimum 8 experiments from the following). <ol style="list-style-type: none"> 1. To determine Planck's constant using Photocell. 2. To determine Planck's constant using LED. 3. To determine wavelength of spectral lines of mercury source using spectrometer. 4. To determine the value of Rydberg's constant using diffraction grating and hydrogen discharge tube. 5. To determine the wavelength of H-alpha emission line of Hydrogen atom. 6. To determine fine structure constant using fine structure separation of sodium D-lines using a plane diffraction grating. 7. To determine the value of e/m by Magnetic focusing or Bar magnet. 8. To determine the ionization potential of mercury. 9. To setup the Millikan oil drop apparatus and determine the charge of an electron. 10. To determine the absorption lines in the rotational spectrum of Iodine vapour. 11. To determine the force constant and vibrational constant for the iodine molecule from its absorption spectrum. 12. To determine the wavelength of laser using diffraction by single slit/double slits. 13. To determine wavelength of He-Ne laser using plane diffraction grating. 14. To determine angular spread of He-Ne laser using plane diffraction grating. 15. Study of Raman scattering by CCl₄ using laser and spectrometer/CDS. <p>NOTE: Students have to perform at-least EIGHT Experiments from the above list.</p>			

Pedagogy: Demonstration/Experiential Learning / Self Directed Learning etc.

Formative Assessment for Practical	
Assessment Occasion/ type	Marks
Practical Internal Assessment	10
Record	10
Attendance	05
Total	25 Marks
<i>Formative Assessment as per UNIVERSITY guidelines are compulsory.</i>	

References	
1	Practical Physics, D.C. Tayal, First Millennium Edition, 2000, Himalaya Publishing House.
2	B.Sc. Practical Physics, C.L. Arora, Revised Edition, 2007, S. Chand & Comp. Ltd.
3	An Advanced Course in Practical Physics, D. Chatopadhyaya, P.C. Rakshith, B. Saha, Revised Edition, 2002, New Central Book Agency Pvt. Ltd.
4	Physics through experiments, B. Saraf, 2013, Vikas Publications.

SEMESTER - VI

Category	Course code	Title of the Paper	Marks			Teaching hours/week			Credit	Duration of exams (Hours)
			IA	SEE	Total	L	T	P		
DSC	21BSC5C5PHY7L	Elements of Condensed Matter Physics and Nuclear Physics	40	60	100	4	-	-	4	2
	21BSC6C6PHY7P	Elements of Condensed Matter Physics and Nuclear Physics (Practical)	25	25	50	-	-	4	2	3
	21BSC6C6PHY8L	Electronic Instrumentation and Statistical Physics	40	60	100	4	-	-	4	2
	21BSC6C6PHY8P	Electronic Instrumentation (Practical)	25	25	50	-	-	4	2	3
Total Marks					300	Semester Credits			12	

Program Name	B. Sc. Physics	Semester	VI
Course Title	Elements of Condensed Matter Physics and Nuclear Physics		
Course Code:	21BSC6C6PHY7L	No. of Credits	04
Contact hours	60 Hours	Duration of SEA/Exam	2 Hours
Formative Assessment Marks	40	Summative Assessment Marks	60

Course Pre-requisite(s):

Course Outcomes (COs): After the successful completion of the course, the student will be able to:

- Explain the basic properties of nucleus and get the idea of its inner information.
- Understand the concepts of binding energy and binding energy per nucleon v/s mass number graph.
- Describe the processes of alpha, beta and gamma decays based on well-established theories.
- Explain the basic aspects of interaction of gamma radiation with matter by photoelectric effect, Compton scattering and pair production.
- Explain the different nuclear radiation detectors such as ionization chamber, Geiger-Mueller counter etc.
- Explain the basic concept of scintillation detectors, photo-multiplier tube and semiconductor detectors.

Contents	60 Hours
<p style="text-align: center;">Unit-I</p> <p>Crystal systems and X-rays: Crystal structure: Space Lattice, Lattice translational vectors, Basis of crystal structure, Types of unit cells, primitive, non-primitive cells. Seven crystal system, Coordination numbers, Miller Indices, Expression for inter planner spacing.</p> <p>X Rays: Production and properties of X rays, Coolidge tube, Continuous and characteristic X-ray spectra; Moseley's law. X-Ray diffraction, Scattering of X-rays, Bragg's law and powder diffraction.</p> <p>Free electron theory of metals: Classical free electron model (Drude-Lorentz model), expression for electrical and thermal conductivity, Weidman-Franz law, Failure of classical free electron theory; Quantum free electron theory, Fermi level and Fermi energy, Fermi-Dirac distribution function (expression for probability distribution $F(E)$, statement only); Fermi Dirac distribution at $T=0$ and $E < E_f$, at $T \neq 0$ and $E > E_f$, $F(E)$ vs E plot at $T = 0$ and $T \neq 0$. Density of states for free electrons (statement only, no derivation). Qualitative discussion of lattice vibration, concept of Phonons and Hall Effect in metals (Qualitative).</p>	15 Hours

<p style="text-align: center;">Unit-II</p> <p>Dielectrics and Superconductivity:</p> <p>Dielectrics: Static dielectric constant, polarizability (electronic, ionic and orientation), calculation of Lorentz field (derivation), Clausius-Mosotti equation (derivation), dielectric loss. Piezo electric effect, cause, examples and applications.</p> <p>Superconductivity: Definition, Experimental results – Zero resistivity and Critical temperature– The critical magnetic field – Meissner effect, Type I and type II superconductors.</p>	15 Hours
<p style="text-align: center;">Unit-III</p> <p>General Properties of Nuclei: Constituents of nucleus and their intrinsic properties, quantitative facts about mass, radii, charge density (matter density), binding energy, main features of binding energy versus mass number curve, angular momentum, parity, magnetic moment, electric moments</p> <p>Radioactivity: Radioactivity: definition of radioactivity, half life, mean life, radioactivity equilibrium (a) Alpha decay: basics of α-decay processes, theory of α emission (brief), Gamow factor, Geiger-Nuttall law. (b) β-decay: energy kinematics for β-decay, positron emission, electron capture, neutrino hypothesis. (c) Gamma decay: Gamma rays emission & kinematics, internal conversion (Definition)</p>	15 Hours
<p style="text-align: center;">Unit-VI</p> <p>Interaction of Nuclear Radiation with Matter: Gamma ray interaction through matter, photoelectric effect, Compton scattering, pair production, Energy loss due to ionization (quantitative description of Bethe Block formula), energy loss of electrons, introduction of Cerenkov radiation</p> <p>Nuclear Radiation Detector: Gas detectors: estimation of electric field, mobility of particle, for ionization chamber and GM Counter. Basic principle of Scintillation Detectors and construction of photo-multiplier tube (PMT). Particle Accelerators: LINAC: Principle, construction, working, expression for energy. Cyclotrons: Principle, construction, working, expression for Cyclotron conditions.</p>	15 Hours

Suggested Activities:

- 1) Students to construct seven crystal systems with bamboo sticks and rubber bands. Use foam ball as atoms and study the BCC and FCC systems.
- 2) Students to search the characteristic X ray wavelength of different atoms/elements and plot characteristic wavelength vs atomic number and analyse the result and draw the inference.
- 3) Magnetic field lines are invisible. Students to trace the magnetic field lines using bar magnet and needle compass. <https://nationalmaglab.org/magnet-academy/try-this-at-home/drawing-magnetic-field-lines/> ,
- 4) Using vegetable oil and iron fillings students to make ferrofluids and see how it behaves in the presence of magnetic field. <https://nationalmaglab.org/magnet-academy/try-this-at-home/making-ferrofluids/>
- 1) Study the decay scheme of selected alpha, beta & gamma radioactive sources with the help of standard nuclear data book.
- 2) Calculate binding energy of some selected light, medium and heavy nuclei. Plot the graph of binding energy versus mass number A
- 3) Study the decay scheme of standard alpha, beta and gamma sources using nuclear data book.
- 4) Make the list of alpha emitters from Uranium series and Thorium series. Search the kinetic energy of alpha particle emitted by these alpha emitters. Collect the required data such as half life or decay constant. Verify Geiger-Nuttall in each series.
- 5) Study the Z dependence of photoelectric effect cross section.
- 6) Study the Z dependence of common cross section for selected gamma energies and selected elements through theoretical calculation.
- 7) List the materials and their properties which are used for photocathode of PMT.
- 8) Study any two types of PMT and their advantages and disadvantages.

Pedagogy: Lecture/ PPT/ Videos/ Animations/ Role Plays/ Think-Pair-Share/ Predict-Observe-Explain/ Demonstration/ Concept mapping/ Case Studies examples/ Tutorial/ Activity/ Flipped Classroom/ Jigsaw/ Field based Learning/ Project Based Learning/ Mini Projects/ Hobby Projects/ Forum Theatre/ Dance/ Problem Based Learning/ Game Based Learning/ Group Discussion/ Collaborative Learning/ Experiential Learning / Self Directed Learning etc.

Formative Assessment for Theory	
Assessment Occasion/ type	Marks
Internal Assessment	10
Project/Assignment	10
Activities	10
Presentation (Oral/Poster)	10
Total	40 Marks
<i>Formative Assessment as per UNIVERSITY guidelines are compulsory.</i>	

References:**Text Books:**

1. R. K. Puri and V.K. Babber, Solid State Physics, 1st Edition 2004, S. Chand publications.
2. B. S. Saxena, P. N. Saxena, Fundamentals of Solid State Physics, 2017, Pragati prakashan Meerut.
3. Kenneth S. Krane, Introductory nuclear Physics, 2008, Wiley India Pvt. Ltd.
4. Irving Kaplan, Nuclear Physics, 1977, Narosa Publishing House.

Reference Books:

1. Charles Kittel, Introduction to Solid State Physics, 1996, VII edition,
5. A. J. Dekker, Solid State Physics, 2000, MacMillan India Ltd,
6. M. A. Wahab, Essential of Crystallography, 2009, Narosa Publications
7. S. O. Pillai, Solid State Physics, 2001, New Age Int. Publishers.
8. Bernard L. Cohen, Concepts of Nuclear Physics, 1998, Tata McGraw Hill,
9. R.A. Dunlap, Introduction to the Physics of Nuclei & Particles, 2004, Thomson Asia.
10. D. H. Perkins, Introduction to High Energy Physics, 2001, Cambridge University, Press.
11. K. Heyde, Basic ideas and concepts in Nuclear Physics - An Introductory Approach, 2004, Institute of Physics (IOP) Publishing.
12. G. F. Knoll, Radiation detection and measurement, 2000, John Wiley & Sons.
13. Syed Naeem Ahmed, Physics and Engineering of Radiation Detection, 2007, Academic Press, Elsevier.

Course Title	Elements of Condensed Matter and Nuclear Physics (Practical)		Practical Credits	02
Course Code	21BSC6C6PHY7P		Contact Hours	04 Hours
Formative Assessment	25 Marks	Summative Assessment		25 Marks
Practical Content				
Lab experiments: (Minimum 8 experiments from the following).				
<ol style="list-style-type: none">1. Determination of Plank's constant by Photo Cell2. Hall Effect in semiconductor: determination of mobility, hall coefficient.3. Energy gap of semiconductor (diode/transistor) by reverse saturation method4. Thermistor energy gap5. Fermi Energy of Copper6. Analysis of X-ray diffraction spectra and calculation of lattice parameter.7. Plank's constant by LED8. Specific Heat of Solid by Electrical Method9. Determination of Dielectric Constant of polar liquid.10. Determination of dipole moment of organic liquid11. B-H Curve Using CRO.12. Spectral Response of Photo Diode and its I-V Characteristics.13. Determination of particle size from XRD pattern using Debye-Scherrer formula.14. Measurement of susceptibility of paramagnetic solution (Quinck's Tube Method).15. Measurement of susceptibility of paramagnetic solid (Gouy's Method)				
NUCLEAR PHYSICS				
<ol style="list-style-type: none">1. Study the characteristics of Geiger-Müller Tube. Determine the threshold voltage, plateau region and operating voltage.2. Study the absorption of beta particles in aluminium foils using GM counter. Determine mass attenuation coefficient of Aluminium foils.3. Study the absorption of beta particles in thin copper foils using G M counter and determine mass attenuation coefficient.4. Study the attenuation of gamma rays in lead foils using Cs-137 source and G M counter. Calculate mass attenuation coefficient of Lead for Gamma.5. Determine the end point energy of Tl-204 source by studying the absorption of beta particles in aluminium foils.6. Study the attenuation of absorption of gamma rays in polymeric materials using Cs-137 source and G M counter.				

Pedagogy: Demonstration/Experiential Learning / Self Directed Learning etc.

Formative Assessment for Practical	
Assessment Occasion/ type	Marks
Practical Internal Assessment	10
Record	10
Attendance	05
Total	25 Marks
<i>Formative Assessment as per UNIVERSITY guidelines are compulsory</i>	

References	
1	B. Saraf et al., Experiment in Physics, 1998, Vikas Publications.
2	S.P. Singh : Advanced Practical Physics, 1976, Pragati Prakashan.
3	Adrian C. Melissinos, Experiments in Modern Physics, 2003, Gulf Professional Publishing.
4	Misra and Misra, Physics Lab. Manual, 2000, South Asian publishers.
5	Gupta and Kumar, Practical physics, 1976, Pragati Prakashan.

Program Name	B. Sc. Physics	Semester	VI
Course Title	Electronic Instrumentation and Statistical Physics		
Course Code:	21BSC6C6PHY8L	No. of Credits	04
Contact hours	60 Hours	Duration of SEA/Exam	2 Hours
Formative Assessment Marks	40	Summative Assessment Marks	60

Course Pre-requisite(s):

Course Outcomes (COs): After the successful completion of the course, the student will be able to:

- Identify different types of tests and measuring instruments used in practice and understand their basic working principles.
- Get hands on training in wiring a circuit, soldering, making a measurement using an electronic circuit used in instrumentation.
- Have an understanding of the basic electronic components viz., resistors, capacitors, inductors, discrete and integrated circuits, colour codes, values and pin diagram, their practical use.
- Understanding of the measurement of voltage, current, resistance value, identification of the terminals of a transistor and ICs.
- Identify and understand the different types of transducers and sensors used in robust and hand-held instruments.
- Understand and give a mathematical treatment of the working of rectifiers, filter, data converters and different types of transducers.
- Connect the concepts learnt in the course to their practical use in daily life.
- Develop basic hands-on skills in the usage of oscilloscopes, multimeters, rectifiers, amplifiers, oscillators and high voltage probes, generators and digital meters.
- Servicing of simple faults of domestic appliances: Iron box, immersion heater, fan, hot plate, battery charger, emergency lamp and the like.

Contents

60 Hours

Unit-I

Power supply: AC power and its characteristics, Single phase and three phase, DC power supply and characteristics and its characteristics, line voltage and frequency, Rectifier bridge,
Filters: Capacitor and inductor filters, L-section and π -section filters, ripple factor, electronic voltage regulators, stabilization factor and voltage regulation using ICs.

Basic electrical measuring instruments

Cathode ray oscilloscope- Block diagram, basic principle, electron beam, CRT features, signal display. Basic elements of digital storage oscilloscopes.
 Basic DC voltmeter for measuring potential difference and Extending Voltmeter range.
 Basic DC ammeter, requirement of a shunt, extending of ammeter ranges.

Topics for self-study:

Average value and RMS value of current, Ripple factor, Average AC input power and DC output power, efficiency of a DC power supply. Multirange voltmeter and ammeter.

12 Hours

15 Hours

<p>Activities 3 Hours</p> <ol style="list-style-type: none"> 1. Design and wire your own DC regulated power supply. Power output: 5 V, 10 V, ± 5 V. Components required: A step down transformer, semiconductor diodes (BY126/127), Inductor, Capacitor, Zener diode or 3-pin voltage regulator or IC. Measure the ripple factor and efficiency at each stage. Tabulate the result. 2. Extend the range of measurement of voltage of a voltmeter (analog or digital) using external component and circuitry. Design your own circuit and report. 3. Measure the characteristics of the signal waveform using a CRO and function generator. Tabulate the frequency and time period. Learn the function of Trigger input in an CRO. 4. Learn to use a Storage Oscilloscope for measuring the characteristics of a repetitive input signal. Convince yourself how signal averaging using Storage CRO improves S/N ratio. 	
<p style="text-align: center;">Unit-II</p> <p>Wave form generators, Filters and Data Conversion & Display: Basic principle of standard AF signal generator: Fixed frequency and variable frequency, AF sine and square wave generator, basic Wein-bridge oscillator using operational amplifier.</p> <p>Passive and active filters. Fundamental theorem of filters, types of filters, Circuitry and Cut-off frequency and frequency response of Passive (RC) and Active (op-amp based) filters: Low pass, high pass and band pass.</p> <p>Digital to Analog (D/A) and Analog to Digital (A/D) converters – A/D converter with pre-amplification and filtering. Digital display systems and Indicators- Classification of displays, Light Emitting Diodes (LED) and Liquid Crystal Display (LCD) – Structure and working.</p> <p style="text-align: right;">12 Hours</p> <p>Activities 3 Hours</p> <ol style="list-style-type: none"> 1. Measure the amplitude and frequency of the different waveforms and tabulate the results. Required instruments: A 10 MHz oscilloscope, Function generators (sine wave and square wave). 2. Explore where signal filtering network is used in real life. Visit a nearby telephone exchange and discuss with the Engineers and technicians. Prepare a report. 3. Explore op-amp which works from a single supply biasing voltage (+15V). Construct an inverting/non-inverting amplifier powered by a single supply voltage instead of dual or bipolar supply voltage. 	15 Hours
<p style="text-align: center;">Unit III</p> <p>Classical Statistics: Macrostate & Microstate, Elementary Concept of Ensemble, Phase Space, Entropy and Thermodynamic Probability, Maxwell-Boltzmann Distribution Law, Partition Function, Thermodynamic Functions of an Ideal Gas, Classical Entropy Expression, Gibbs Paradox, Sackur Tetrode equation, Law of Equipartition of Energy (with proof) –Applications to Specific Heat and its Limitations.</p> <p style="text-align: right;">12 Hours</p> <p>Activities: 03 Hours</p> <ol style="list-style-type: none"> 1. Students to learn Binomial distribution using coins and plot the distribution curve. 2. Students to learn Normal distribution using nails and plot the distribution curve. 	15 Hours

Unit-IV		15 Hours
Quantum Statistics: Bose-Einstein distribution law, Thermodynamic functions of a strongly Degenerate Bose Gas, Bose Einstein condensation, Radiation as a photon gas and Thermodynamic functions of photon gas. Bose derivation of Planck's law. Fermi-Dirac Distribution Law, Thermodynamic functions of a Completely and strongly Degenerate Fermi Gas, Fermi Energy, Electron gas in a Metal, Specific Heat of Metals, Relativistic Fermi gas, White Dwarf Stars, Chandrasekhar Mass Limit.		
12 Hours		
Activities:		
03 Hours		
1. Students to list particles belonging to Bosons and Fermions, and understand their characteristics. Students also make the presentation of the study.		
2. Students to search the contribution of Indian scientist J.C. Bose in Bose-Einstein condensation. Students also make the presentation of the study.		

Pedagogy: Lecture/ PPT/ Videos/ Animations/ Role Plays/ Think-Pair-Share/ Predict-Observe-Explain/ Demonstration/ Concept mapping/ Case Studies examples/ Tutorial/ Activity/ Flipped Classroom/ Jigsaw/ Field based Learning/ Project Based Learning/ Mini Projects/ Hobby Projects/ Forum Theatre/ Dance/ Problem Based Learning/ Game Based Learning/ Group Discussion/ Collaborative Learning/ Experiential Learning / Self Directed Learning etc.

Formative Assessment for Theory	
Assessment Occasion/ type	Marks
Internal Assessment	10
Project/Assignment	10
Activities	10
Presentation (Oral/Poster)	10
Total	40 Marks
<i>Formative Assessment as per UNIVERSITY guidelines are compulsory.</i>	

References
1. Physics for Degree students (Third Year) – C.L. Arora and P.S. Hemne, 2014, S, Chand and Co. Pvt. Ltd. (For Unit-1, Power supplies) 2. Electronic Instrumentation, 3 rd Edition, H.S. Kalsi, 2011, McGraw Hill Education India Pvt. Ltd. (For rest of the syllabus) 3. Instrumentation – Devices and Systems (2 nd Edition)– C.S. Rangan, G.R. Sarma, V.S.V. Mani, 2009, Tata McGraw Hill Education Pvt. Ltd. (Especially for circuitry and analysis of signal generators and filters). 4. Statistical Physics, Berkeley Physics Course, F. Reif, 2008, Tata McGraw-Hill 5. Statistical and Thermal Physics, S. Lokanathan and R.S. Gambhir. 1991, Prentice Hall 6. An Introduction to Statistical Mechanics & Thermodynamics, R.H. Swendsen, 2012, Oxford University Press.

Course Title	Electronic Instrumentation (Practical)	Practical Credits	2
Course Code	21BSC6C6PHY8P	Contact Hours	4 Hours
Formative Assessment	25 Marks	Summative Assessment	25 Marks

Practical Content

List of experiments (At least 8 experiments to be performed)

1. Construct a DC power supply using a bridge rectifier and a capacitor filter. Use a Zener diode or a 3-pin voltage regulator and study the load and line regulation characteristics. Measure ripple factor with and without filter and compare with theoretical values.
2. Calibration of a low range voltmeter using a potentiometer
3. Calibration of an ammeter using a potentiometer
4. Design and construct a Wien bridge oscillator (sine wave oscillator) using μA 741 op-amp. Choose the values of R and C for a sine wave frequency of 1 KHz. Vary the value of R and C to change the oscillation frequency.
5. Design and construct a square wave generator using μA 741 op-amp. Determine its frequency and compare with the theoretical value. Also measure the slew rate of the op-amp. If the 741 is replaced by LM318, study how does the waveform compare with the previous one.
6. Study the frequency response of a first order op-amp low pass filter
7. Study the frequency response of a first order op-amp low pass filter
8. Study the characteristics of *pn*-junction of a solar cell and determine its efficiency.
9. Study the illumination intensity of a solar cell using a standard photo detector (e.g., lux meter).
10. Study the characteristics of a LED (variation of intensity of emitted light).
11. Study the characteristics of a thermistor (temperature coefficient of resistance)
12. Study the characteristics of a photo-diode
13. Determine the coupling coefficient of a piezo-electric crystal.
14. Study the amplitude modulation using a transistor.
15. Performance analysis of A/D and D/A converter using resistor ladder network and op-amp.

Pedagogy: Lecture/ PPT/ Videos/ Animations/ Role Plays/ Think-Pair-Share/ Predict-Observe-Explain/ Demonstration/ CGroup Discussion/ Collaborative Learning/ Experiential Learning / Self Directed Learning etc.

Formative Assessment for Practical	
Assessment Occasion/ type	Marks
Practical Internal Assessment	10
Record	10
Attendance	05
Total	25 Marks
<i>Formative Assessment as per University guidelines are compulsory</i>	
References	
1. B. L. Flint and H.T. Worsnop, Advanced Practical Physics for students,1971, Asia Publishing House. 2. C. L. Arora, B.Sc. Practical Physics, (Revised Edition), 2007, S. Chand and Co. Ltd. 3. D. C. Tayal, Practical Physics, First Millennium Edition, 2000, Himalaya Publishing House,	

Employability and skill development

The whole syllabus is prepared with a focus on employability.

Skill development achieved: Fundamental understanding of the working of test and measuring instruments. Operating and using them for measurements. Servicing of laboratory equipment for simple cable faults, loose contacts and discontinuity.

Job opportunities: Lab Assistant/Scientific Assistant in hospitals, R and D institutions, educational institutions.


Chairman
Board of Studies
Department of Physics
Davangere University
Davangere, Davangere-07

B. Sc. QUESTION PAPER PATTERN

B.Sc. _____ Semester Degree Examination; December, 2024
(CBCS Scheme; New Syllabus: 2023-24)

SUBJECT: _____ Paper: _____ Paper Code: _____
Time: 3 Hours

Max. Marks: 60

PART-A

1. Answer any **FIVE** questions.

5×2=10M

- 1.
- 2.
- 3.
- 4.
- 5.
- 6.
- 7.
- 8.

Note: Minimum Two questions from each unit.

PART-B

II. Answer any **FIVE** of the following questions.

5×4=20 M

- 9.
- 10.
- 11.
- 12.
- 13.
- 14.
- 15.
- 16.

Note: Minimum Two questions from each unit.

PART-C

III. Answer any **Three** of the following questions.

3×10=30 M

[Each question should contain two sub questions (a) and (b)]

- 17.
- 18.
- 19.
- 20.
- 21.

Note: Minimum one question from each unit.



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Registrar
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